

Spelling list: GCSE Chemistry: Electrolysis 2

Key words for GCSE Chemistry

halide	A halide is an atom from group 7 on the periodic table (halogens) which has gained an electron to form an item.
reactive	If a substance is reactive , it reacts easily.
reactivity	Reactivity refers to how reactive something is.
pure	If a substance is pure , it is not mixed with anything else.
purify	If you purify a substance, it means your turn it into a pure element or compound.
graphite	Graphite is a carbon allotrope.
electrolysis	Electrolysis describes the process of breaking down a compound using an electrical current.
discharged	Discharged means the charge is removed or released.
compound	Two or more elements in a substance make a compound .
ore	A mineral or rock that metal is extracted from is called ore .
inert	If a substance is inert , it means it is unreactive.
subatomic	Protons, neutrons, and electrons are the subatomic particles found in an atom.
allotrope	An allotrope is a form in which an element can exist, for example charcoal, graphite, and diamonds are are allotropes of carbon.
ion	An ion is formed when an atom or a group of atoms gains or loses and electron.